

INVENTOR:
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EXPRESS NO.:
SHEET 1 OF 6

DiFoggio
A Method and Apparatus for Downhole Quantification...
G. Michael Roebuck TELEPHONE NO.: 713-266-1130
EV369817905US DOCKET NO.: 584-30872-US

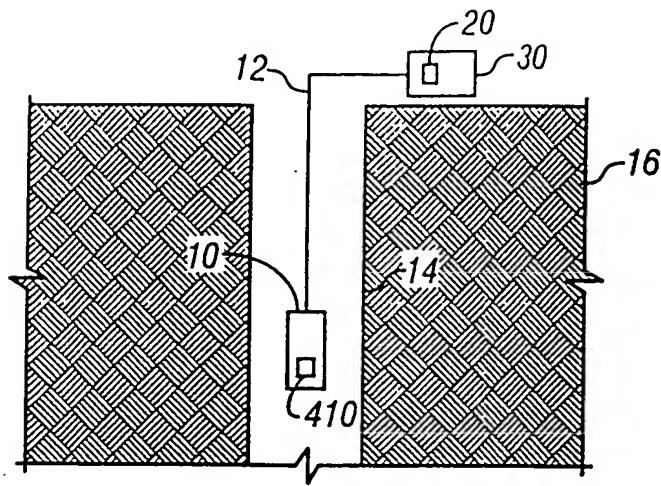


FIG. 1

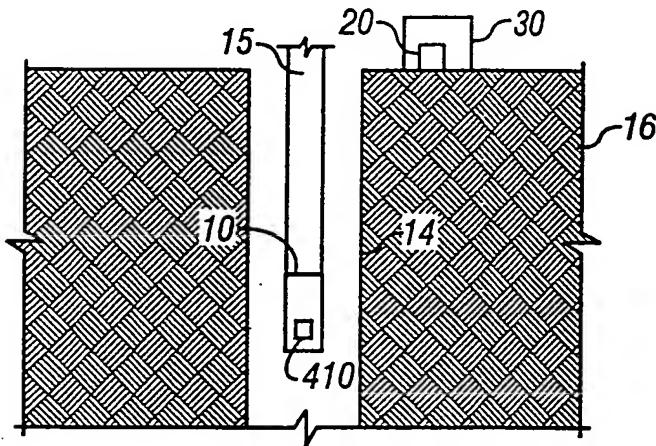


FIG. 2

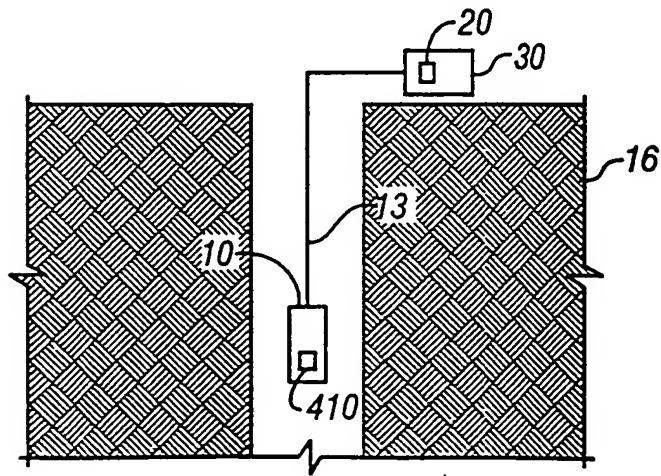


FIG. 3

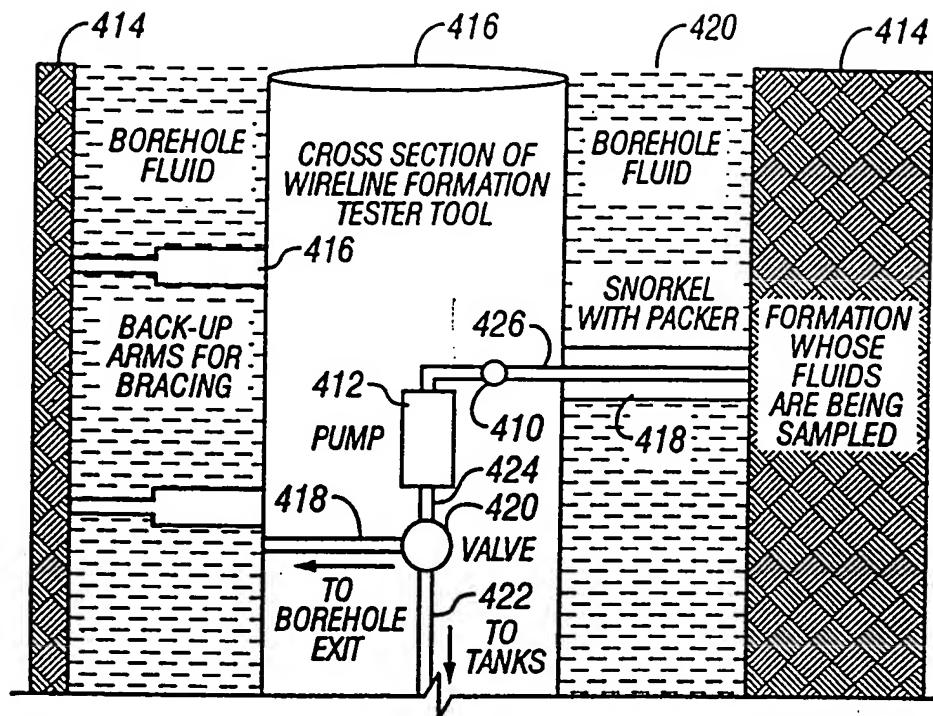


FIG. 4

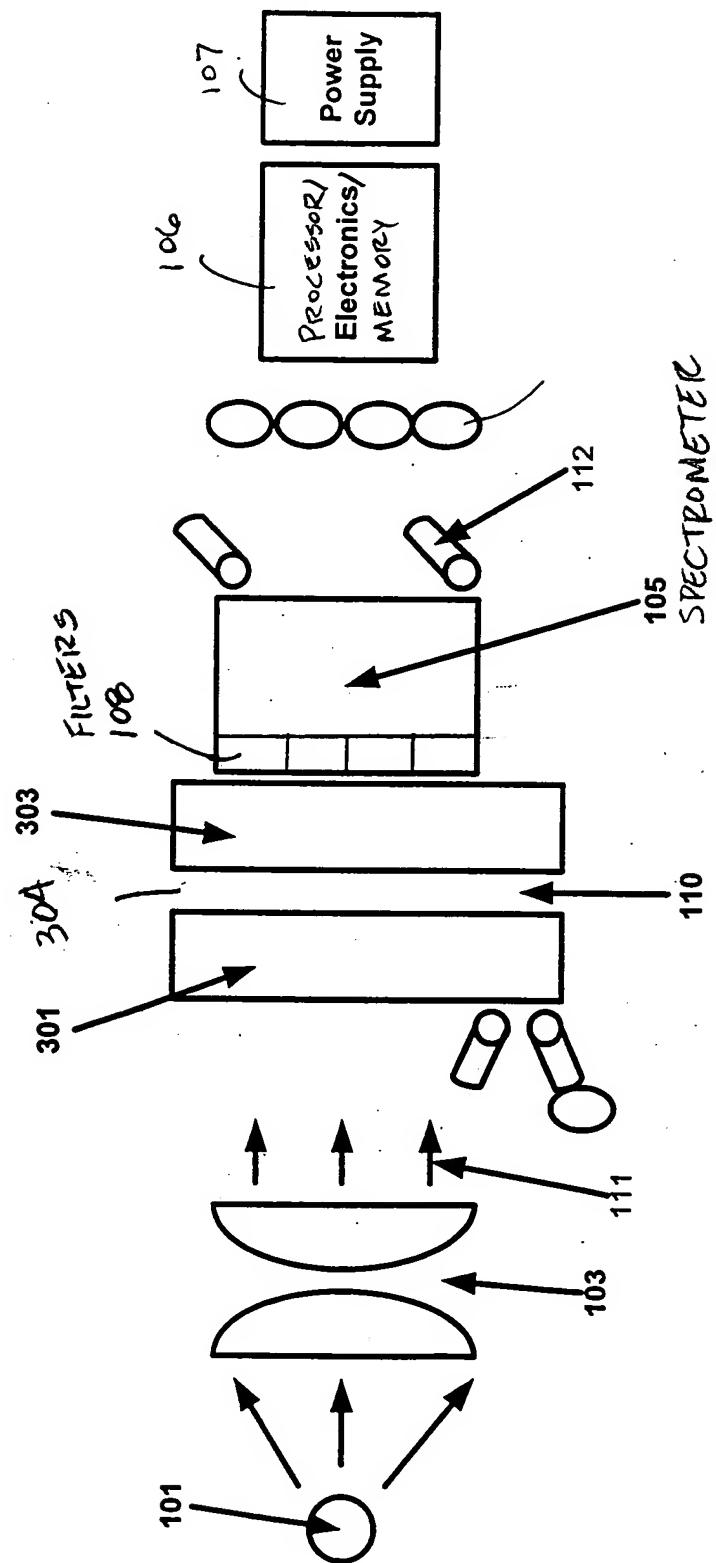


FIG. 5

**Equations Correlating Weight Fraction Methane
in Mixtures of Crude Oil and Methane
to Optical Absorbance and Temperature**

FIG. 6

Methane Weight Fraction = $METHWTF = B0 + B1 * Var1 + B2 * Var2 + B3 * Var3 + B4 * Var4$...

Regression Summary for Dependent Variable: **METHWTF**
R=.98093203 R²= .96222765 Adjusted R²= .96151158
F(4,211)=1343.8 p<0.00000 Std.Error of estimate: .04992

B

0.06514 = B0 = Intercept

11.1756 = B1

0.00087 = B2

-2.66167 = B3

2.63245 = B4

Var1 = SQ70_82

Var2 = TEMP_C

Var3 = SRSA1682

Var4 = SRSA1670

SC70-82 = SQUARE(Absorbance_at_1670_nm - Absorbance_at_1682_nm)
SRSA1670 = SQRT(Absorbance_at_1670_nm)
SRSA1682 = SQRT(Absorbance_at_1682_nm)

TEMP_C = Temperature in Degrees Centigrade

TEMP_SQR = Square of Temperature in Degrees C

Equation for Density of Methane [g/cc] as
a Function of Pressure and Temperature
from 100 - 30,000 psia and 75 - 200 C
is fitted by

B
2.771E-03 = Intercept
2.480E-05
-1.120E-09 for Pressure in psi
P²
P³
1.808E-14
T²
-1.308E-07 for Temperature in C
(P/T)³
(P/T)²
1.455E-03
-4.922E-06
5.934E-09

Equation for Optical Absorbance per mm of
Methane as Function of Density and Wavelength
at 11 nm FWHM, Center λ, range of 1668-1684 nm,
for 100-30,000 psia, and 75 - 200 C,
is fitted by

B
-19.9061 = Intercept
Methane Density 0.7747 for Density in g/cc
WaveNumber/1000 3.3326

Regression Summary for Dependent Variable: **METHWTF**
R=.98190316 R²= .96413381 Adjusted R²= .96327986
F(5,210)=1129.0 p<0.00000 Std.Error of estimate: .04876

B

0.03143 = B0 = Intercept

2.53111 = B1

-2.5576 = B2

11.9135 = B3

0.0019 = B4

-6.2E-06 = B5

Equation's Relating Gas Oil Ratio, **GOR**, to Weight Fraction of Methane, **f_m**, and Stock Tank Density, **p_o**, of Oil
1 bbl = 0.159 m³-5.615 cu ft=42 U.S. gal
1 Standard Cubic Foot (SCF) of Methane Gas at 14.7 psia & 60°F is 0.042358 lbs = 19.21327 grams.
Density of Methane at 60 F and 14.7 psia is 0.0006787 gr/cc = 0.042358 lb/ft³.

Letting **V** = Volume, **W** = Weight, **p** = Density, and using subscripts **M** for Methane and **O** for Oil,
GOR = $W_M / (SCF) / V_{Oil} [bbls] = (W_M / (19.21 g/SCF)) / ((W_O / p_O) (1 bbl / 158 983 cc))$
Letting **f_m** = Weight Fraction of Methane,

GOR = $8274.62 p_O / (1 / f_m - 1)$
 $f_m = W_M / (W_M + W_O) = p_W V_M / (p_M V_M + p_O V_O)$ so $W_O = W_M / (1 / f_m - 1)$ which substitutes into above.

$f_m = 1 / (1 + 8274.62 * p_o / GOR)$ where **W_G** and **W_O** are in grams, **p_o** is in g/cc, and **f_m** = Wt. Frac. of Methane

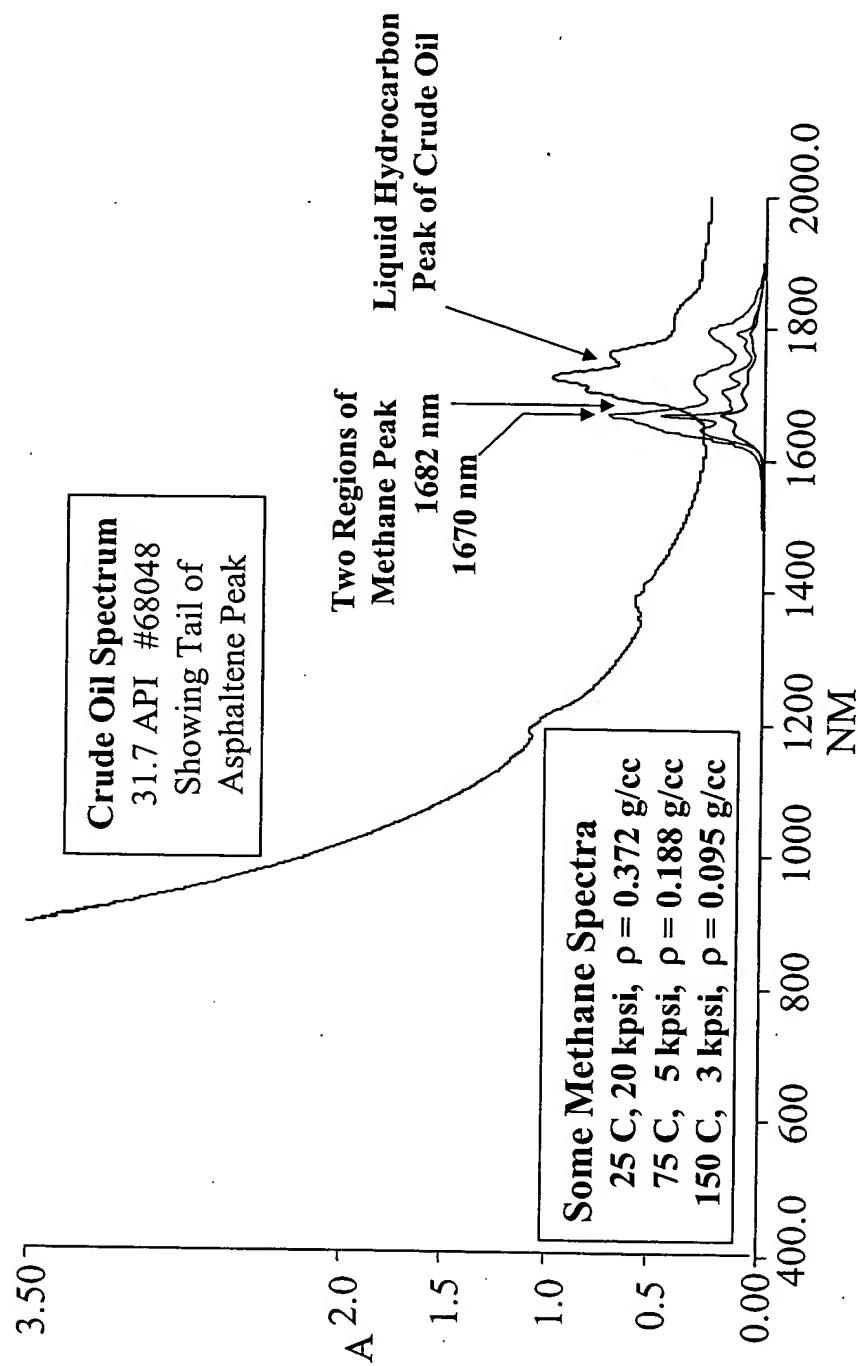


Figure 7

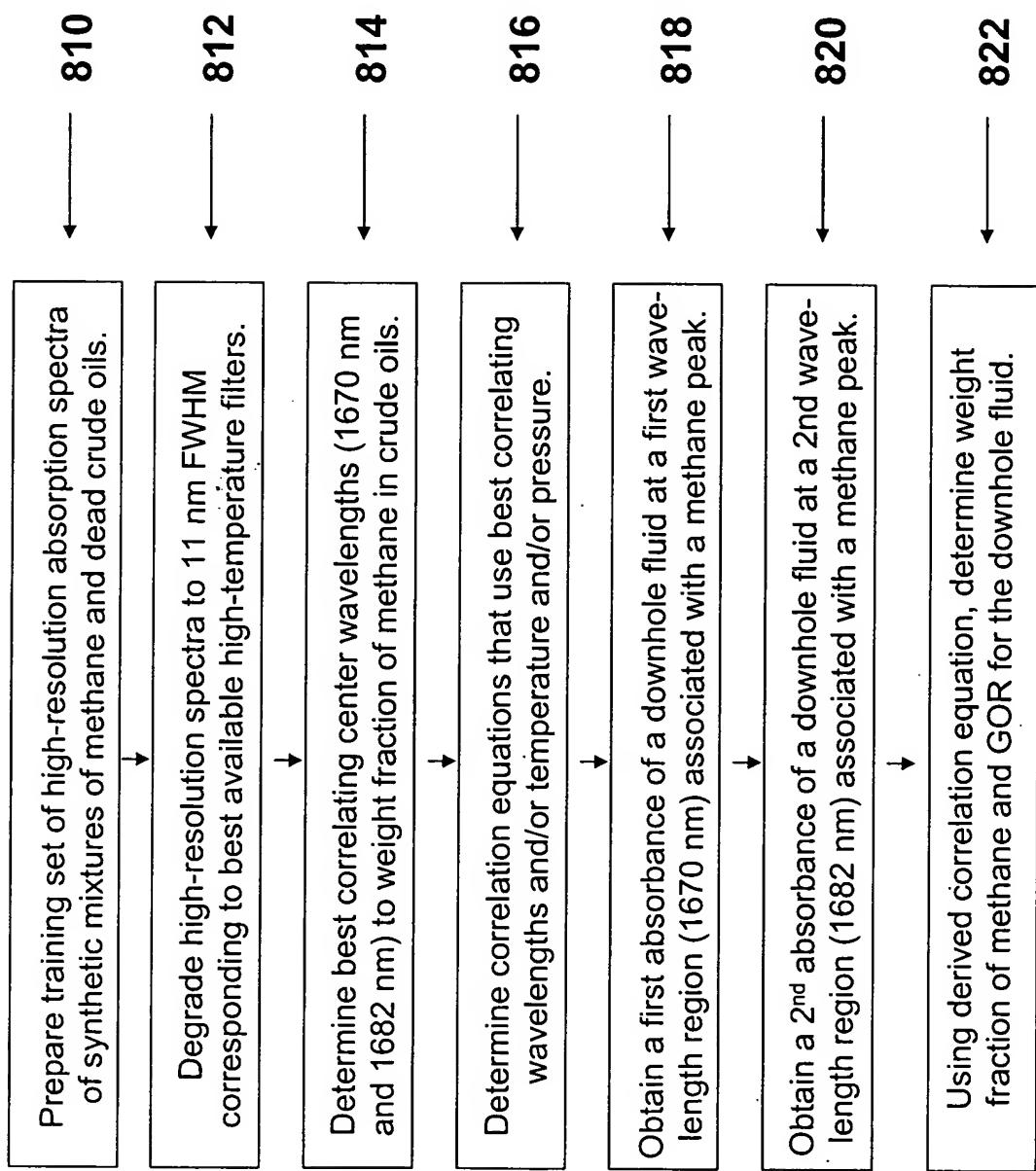


Figure 8